

Probing single molecules with the AFM: Force, dynamics, and function

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1952



Erwin Schrödinger wrote that we would never experiment with just one electron, one atom, or one molecule

E. Schrödinger, *Br. J. Philos. Sci.* **1952**, 3, 233

1959



Richard Feynman told us that there are no physical limitations to maneuvering things atom by atom

There's Plenty of Room at the Bottom
An Invitation to Enter a New Field of Physics

December 29th 1959, annual meeting of the American Physical Society at Caltech

1981: Invention of the STM

G. Binnig, H. Rohrer, Ch. Gerber, E. Weibel
App. Phys. Lett. **1982**, 40, 178



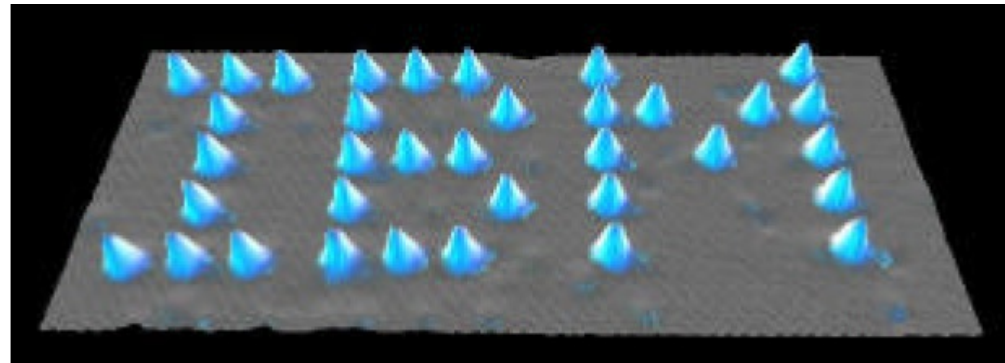
Photos: Copyright © The Nobel Foundation

1986: Binnig, Quate, and Gerber proposed the AFM

Phys. Rev. Lett. **1986**, 59, 178, 930

1990: Positioning single atoms with the STM

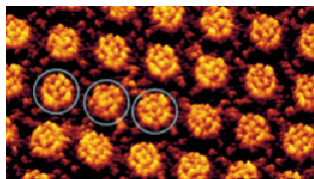
D. M. Eigler, E. K. Schweizer,
Nature **1990**, 344, 524



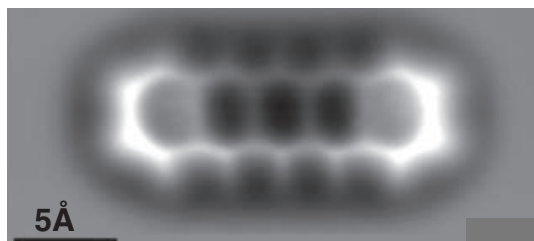
AFM



AFM as a microscope:
Imaging & Characterization

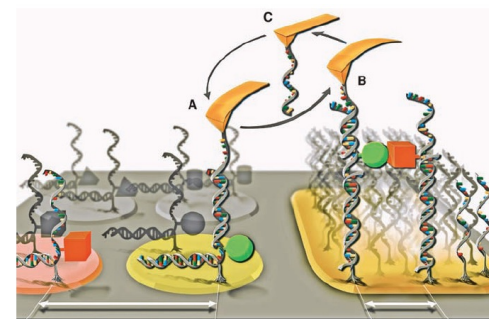


Science 2003, 302, 1002



Science 2009, 325, 1110

AFM as a tool:
Manipulation of molecules

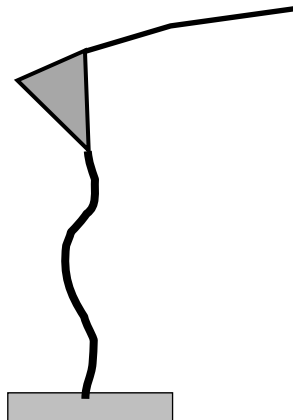


Science 2008, 319, 594

Single molecule force spectroscopy

SCIENCE • VOL. 275 • 28 FEBRUARY 1997

1295

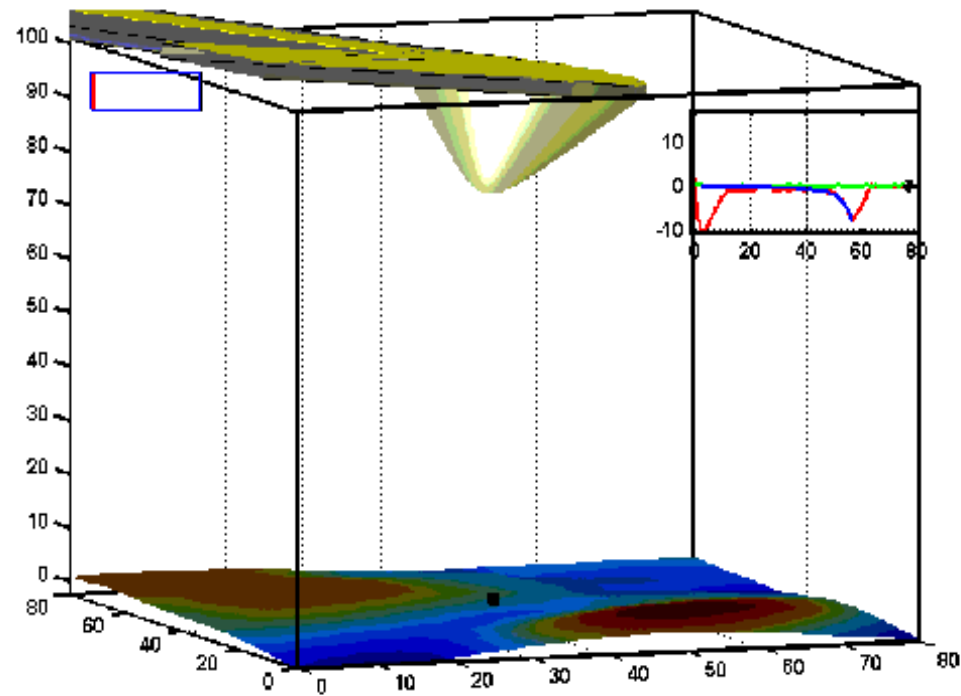


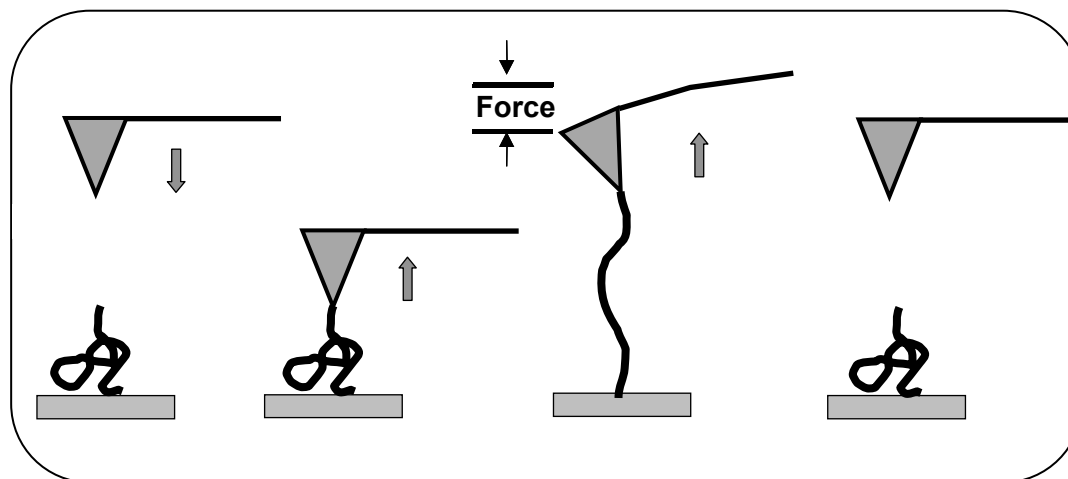
Single Molecule Force Spectroscopy on Polysaccharides by Atomic Force Microscopy

Matthias Rief, Filipp Oesterhelt, Berthold Heymann,
Hermann E. Gaub

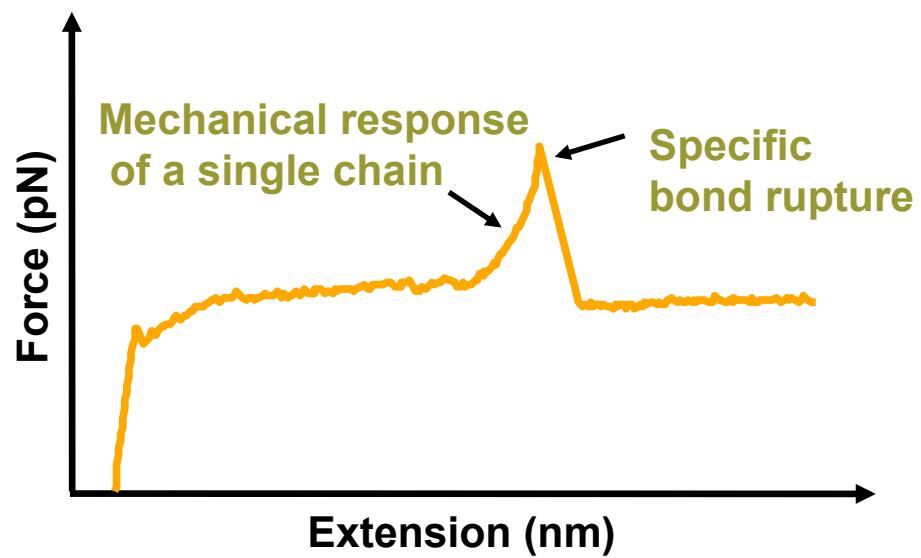
Recent developments in piconewton instrumentation allow the manipulation of single molecules and measurements of intermolecular as well as intramolecular forces. Dextran filaments linked to a gold surface were probed with the atomic force microscope tip by vertical stretching. At low forces the deformation of dextran was found to be dominated by entropic forces and can be described by the Langevin function with a 6 angstrom Kuhn length. At elevated forces the strand elongation was governed by a twist of bond angles. At higher forces the dextran filaments underwent a distinct conformational change. The polymer stiffened and the segment elasticity was dominated by the bending of bond angles. The conformational change was found to be reversible and was corroborated by molecular dynamics calculations.

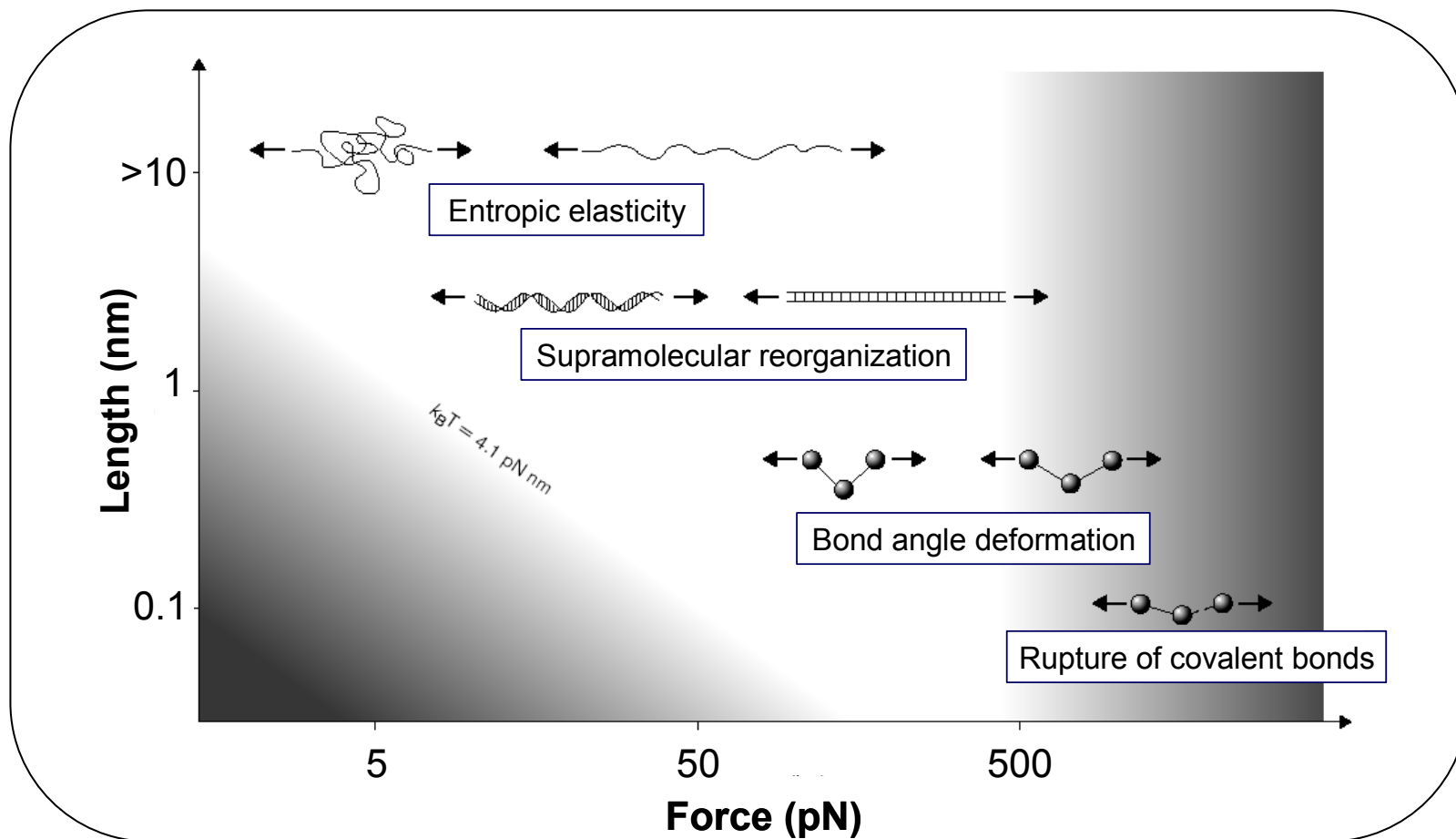
Single molecule force spectroscopy





In solution





From H. E. Gaub et al.
Current Opinion in Chemical Biology **2000**, 4, 524

! It's always the weakest link that breaks

→ The interaction of interest must be the weakest link of the system

→ The design of the tip/molecular species/surface is of prime importance

Why manipulating single entities?

Miniaturization

Ultimate limit ?

A day will come when a tiny piece
of matter will be too small to
fabricate a device or a machine

Single molecule/atom constructions

A single molecule =
an ultra-miniaturized
device/machine

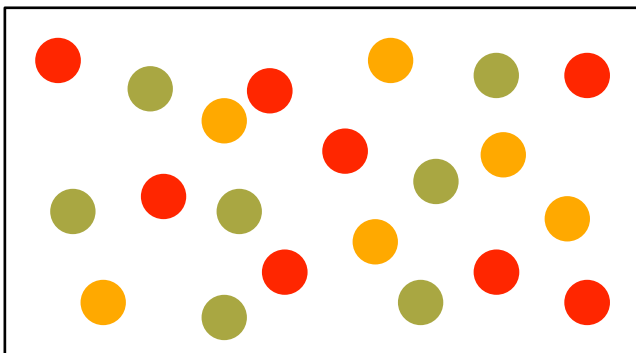


To advance the frontier of the
control of individual molecular
architectures on surfaces

Exploration of single entities

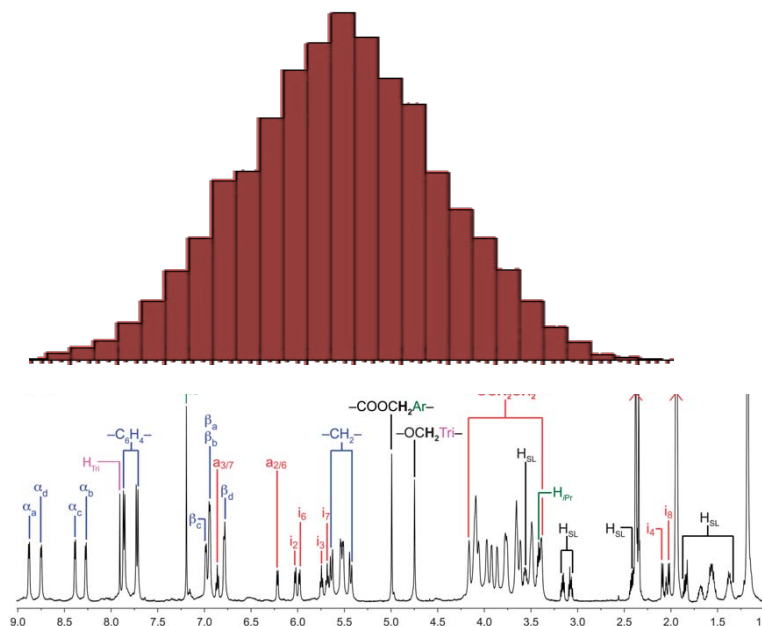
Do the phenomena observed at this scale obey the laws we know for ensemble of species or will force us to rethink our understanding of physics and chemistry?

Classical chemistry

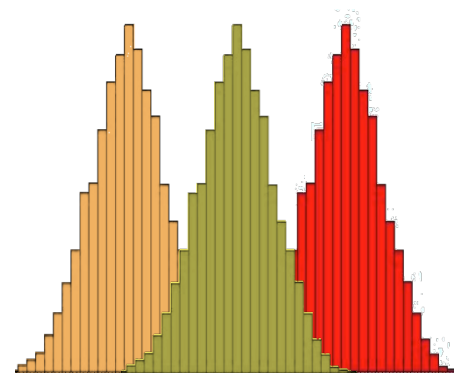
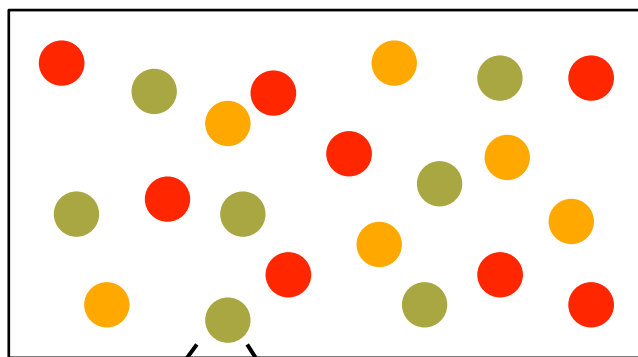


Unit: mole

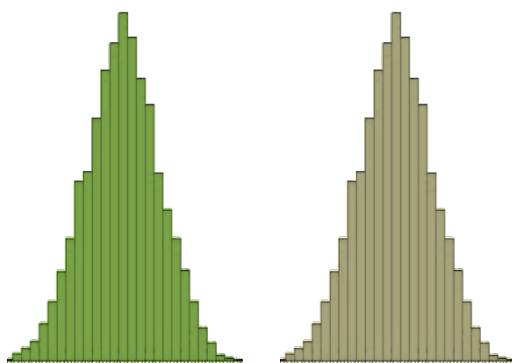
Average behaviour on
billions of molecules



Exploration of single entities



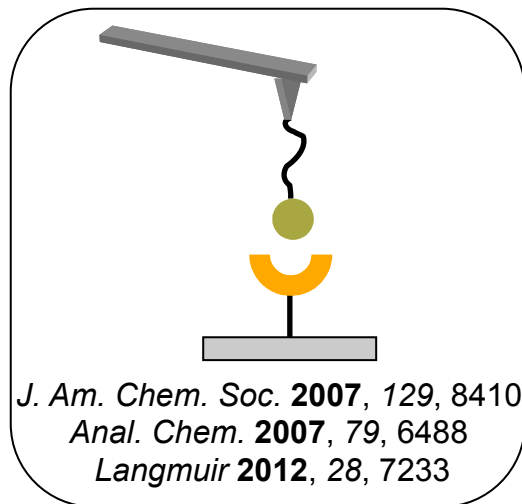
Sub-groups of molecules



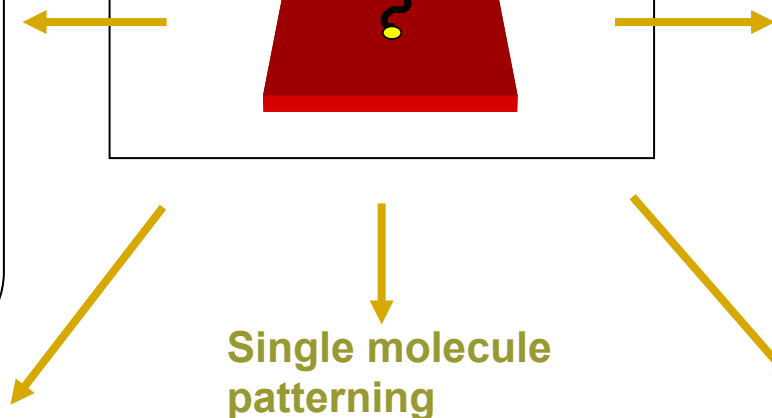
Different behaviour of the same molecule

Unit: 1 single molecule

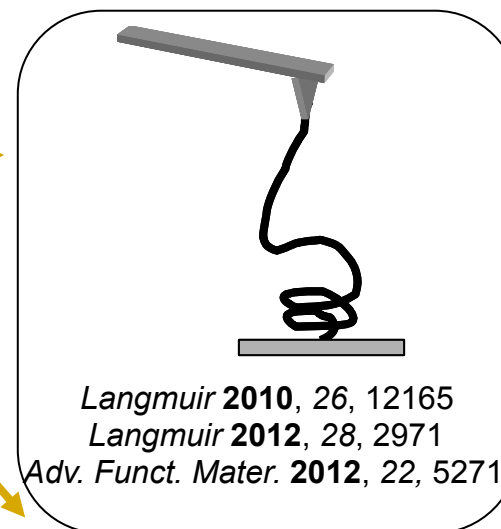
Molecular interactions



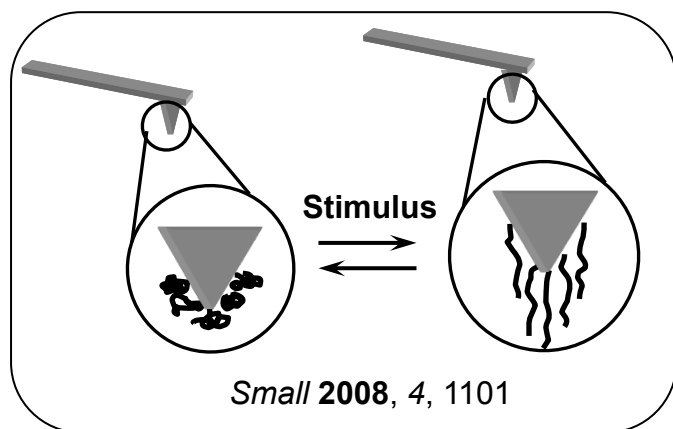
Single molecule force spectroscopy



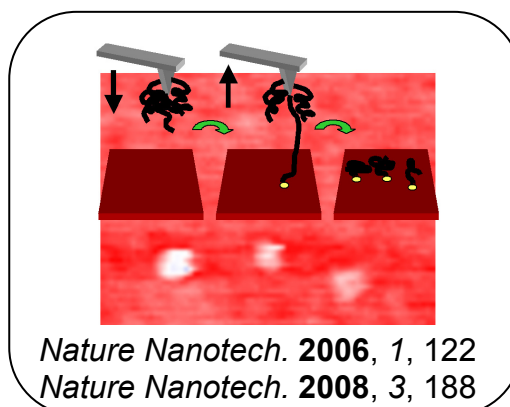
Molecular elasticity and Interactions with surfaces



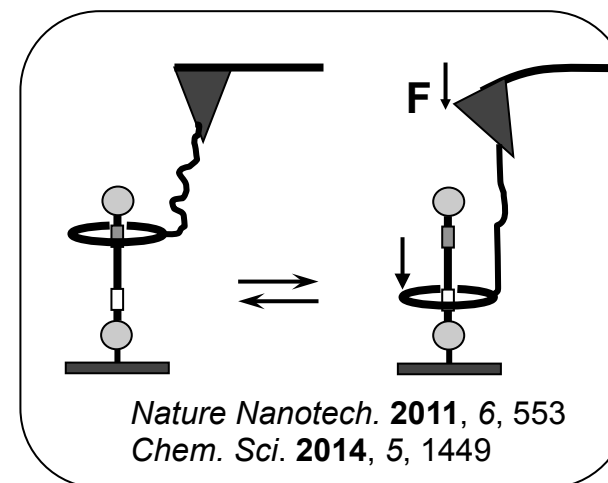
Molecular sensors



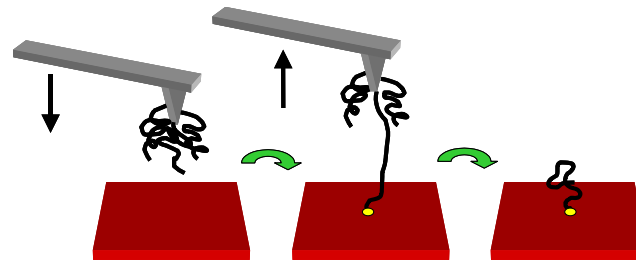
Single molecule patterning



Molecular machines



- **Single molecule patterning**

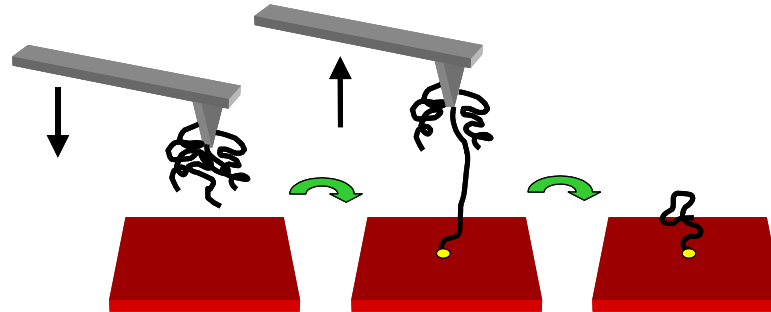


- Molecular interactions
- Molecular machines

→ build complex functional structures molecule by molecule

Three main challenges:

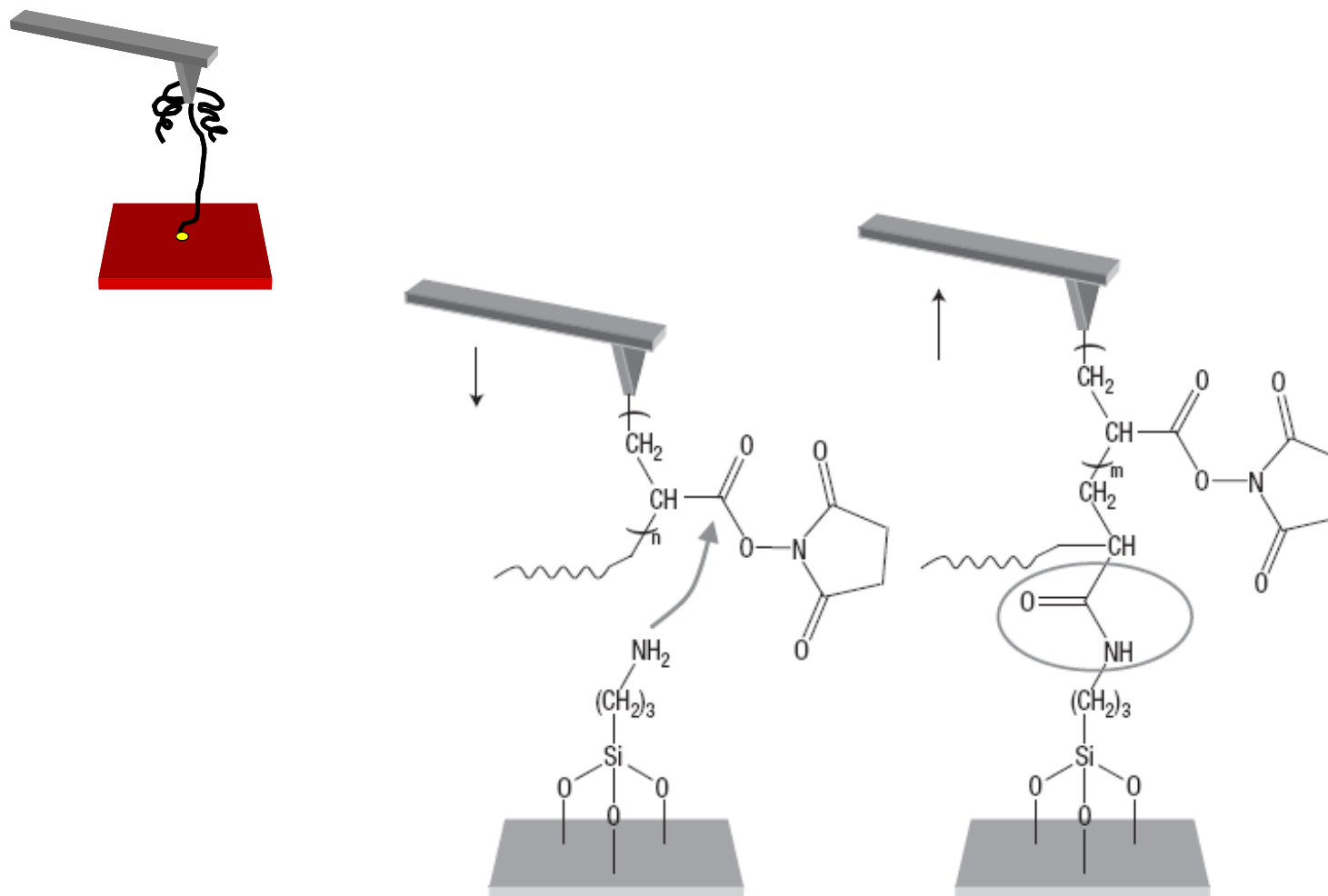
- to position every single molecule in the right place
- to make the molecule form bonds as required
- and to achieve high-throughput fabrication



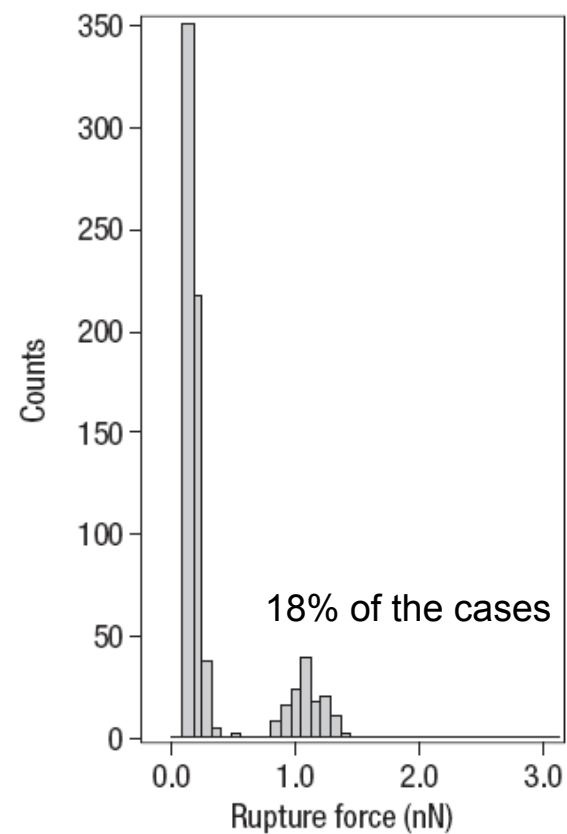
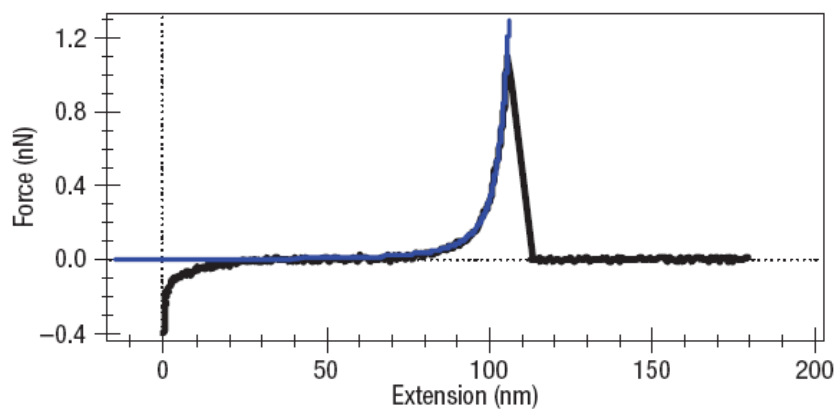
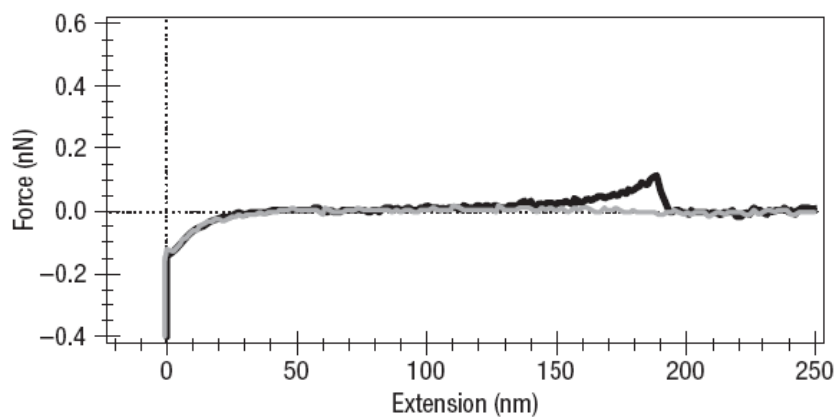
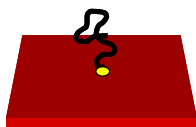
The AFM can deliver and immobilize single molecules, one at a time, on a surface during a single molecule force spectroscopy experiment

Strategy:

- Use AFM tips as reservoirs of single molecules
- The molecules can be linked to a substrate by a chemical reaction
- When the AFM tip is pulled away from the surface, the resulting mechanical force causes the bond between the tip and molecule (the weakest one) to break

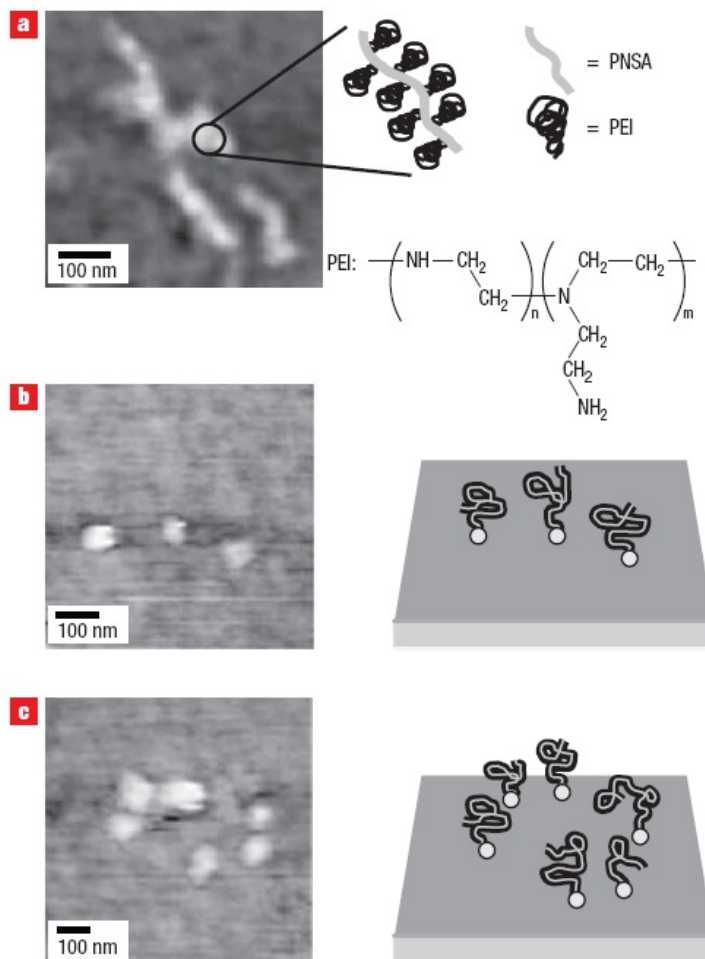


Selective breaking of chemical bonds



Targeted Delivery of Single Molecules

Original deposited chains are post-decorated by a branched polymer to increase their diameter and make them visible



Major disadvantage:

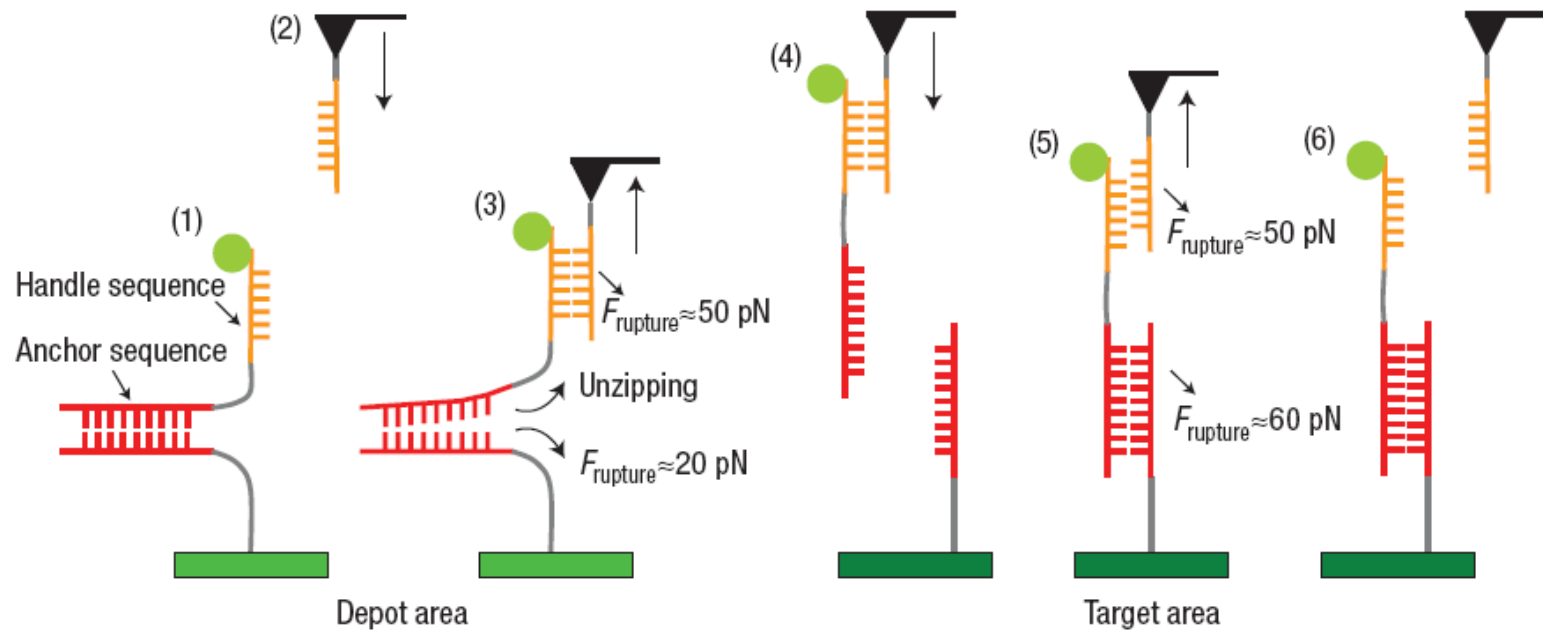
The tip serves as a reservoir: a limited number of molecules can be deposited

→ High-throughput impossible

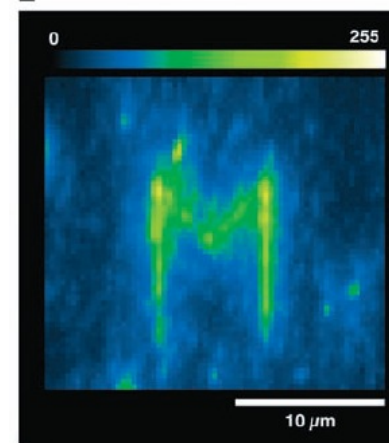
Single-Molecule Cut-and-Paste Surface Assembly

S. K. Kufer,¹ E. M. Puchner,¹ H. Gump, ¹ T. Liedl,² H. E. Gaub¹

We introduce a method for the bottom-up assembly of biomolecular structures that combines the precision of the atomic force microscope (AFM) with the selectivity of DNA hybridization. Functional units coupled to DNA oligomers were picked up from a depot area by means of a complementary DNA strand bound to an AFM tip. These units were transferred to and deposited on a target area to create basic geometrical structures, assembled from units with different functions. Each of these cut-and-paste events was characterized by single-molecule force spectroscopy and single-molecule fluorescence microscopy. Transport and deposition of more than 5000 units were achieved, with less than 10% loss in transfer efficiency.



Gaub and co-workers used their molecular cranes to transport fluorophores and biotin molecules to the target site and, as a proof of concept, they wrote the letter M with 400 fluorescently labelled molecules



- **A single AFM tip can transport and deliver more than 5 000 units, one by one**

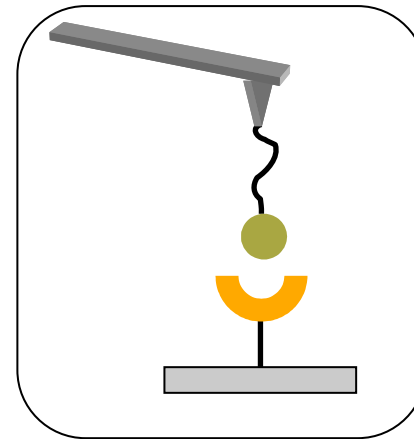
→ Significant step towards high-throughput

Drawbacks:

- Precision of the delivery: ~10 nm
- Some issues still need to be resolved:
 - the AFM tip picks up just 1 unit only 35% of the time,
→ more likely to pick up zero, two or more units
- DNA oligomer used as a 'handle molecule' to transport objects
→ complicated chemical transformations to attach the object on DNA

- Single molecule patterning

- **Molecular interactions**



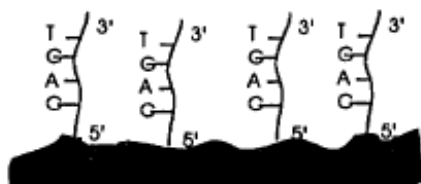
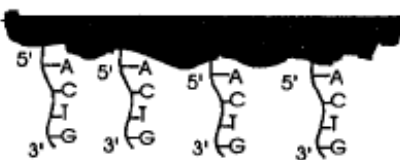
- Molecular machines

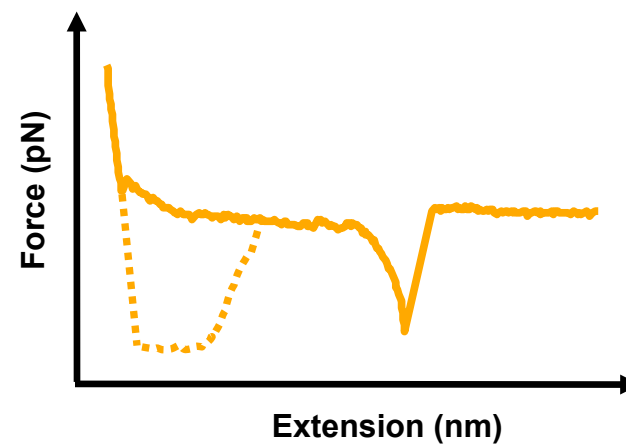
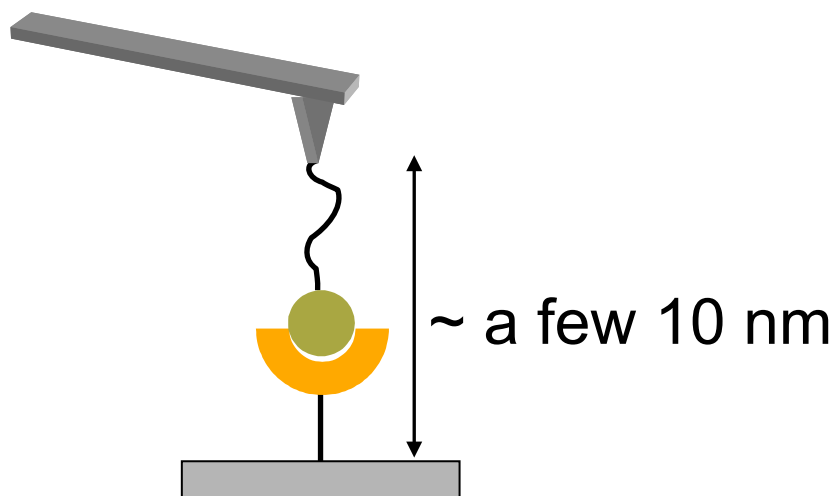
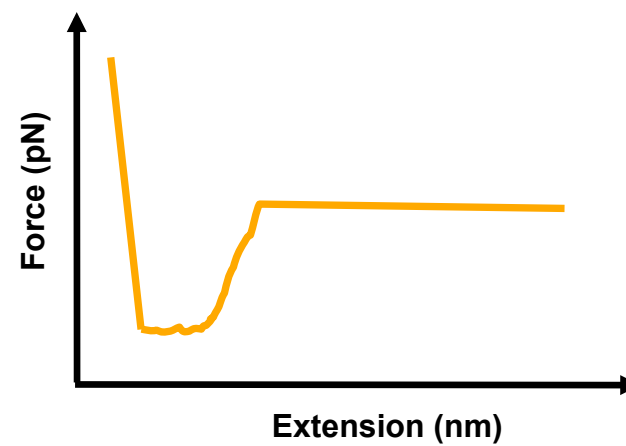
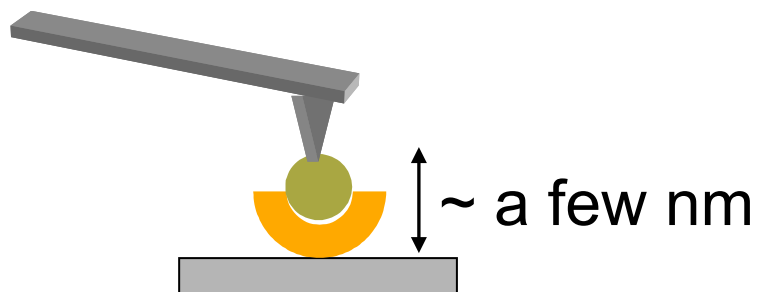


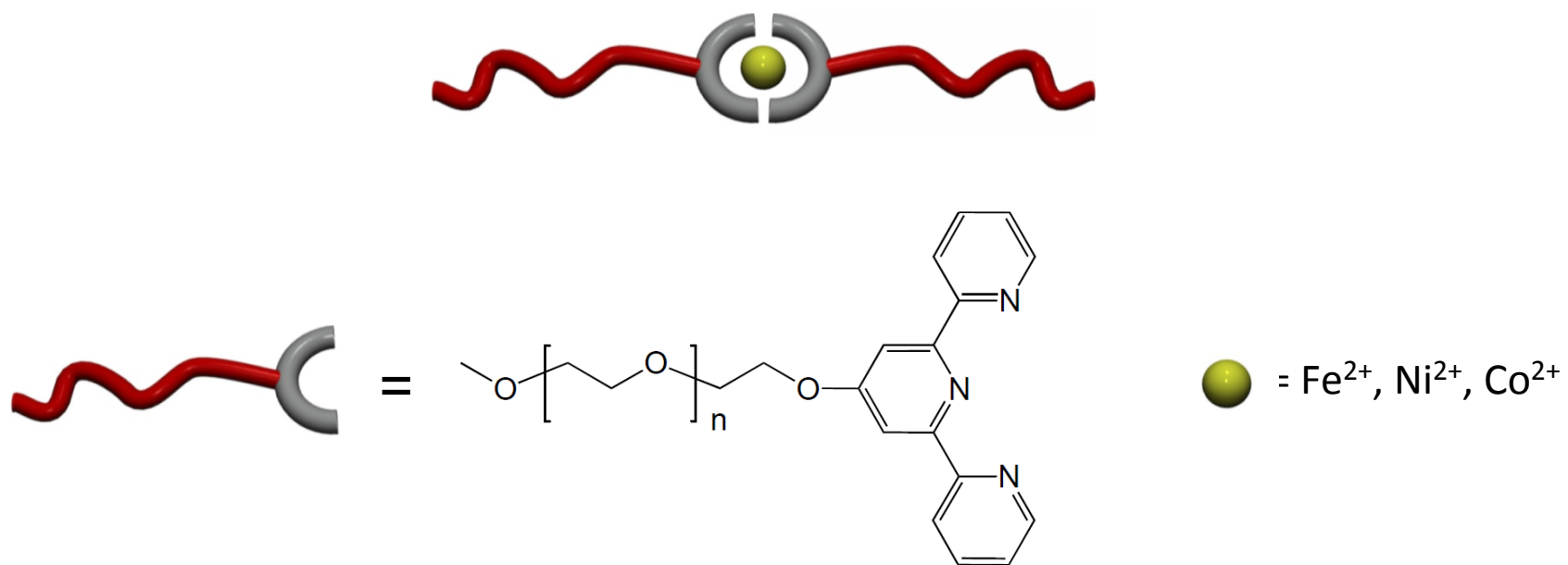
Direct Measurement of the Forces Between Complementary Strands of DNA

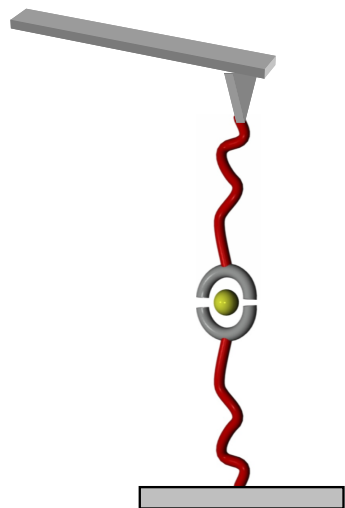
Gil U Lee,* Linda A. Chrisey, Richard J. Colton

Interaction forces between single strands of DNA were measured with the atomic force microscope by a procedure in which DNA oligonucleotides were covalently attached to a spherical probe and surface. Adhesive forces measured between complementary 20-base strands fell into three distinct distributions centered at 1.52, 1.11, and 0.83 nano-newtons, which are associated with the rupture of the interchain interaction between a single pair of molecules involving 20, 16, and 12 base pairs, respectively. When a third long DNA molecule was coupled between complementary surfaces, both intra- and interchain forces were observed. The intrachain interaction resulting from the molecule's elasticity manifested itself as a long-range cohesive force.









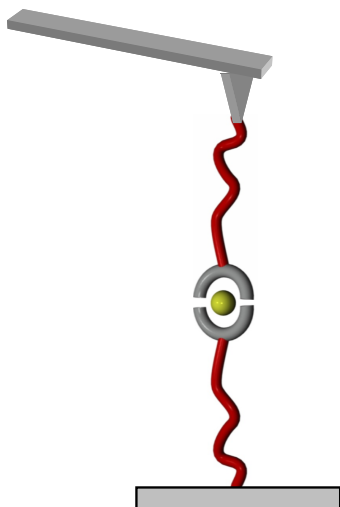
	Force (pN) in water (8 nN s ⁻¹)
PEO-Ni-PEO	64 ± 10
PEO-Co-PEO	45 ± 8
PEO-Fe-PEO	38 ± 8

Thermodynamic binding constants in water

Terp-Ni²⁺ 5.0 x 10¹⁰ M⁻¹

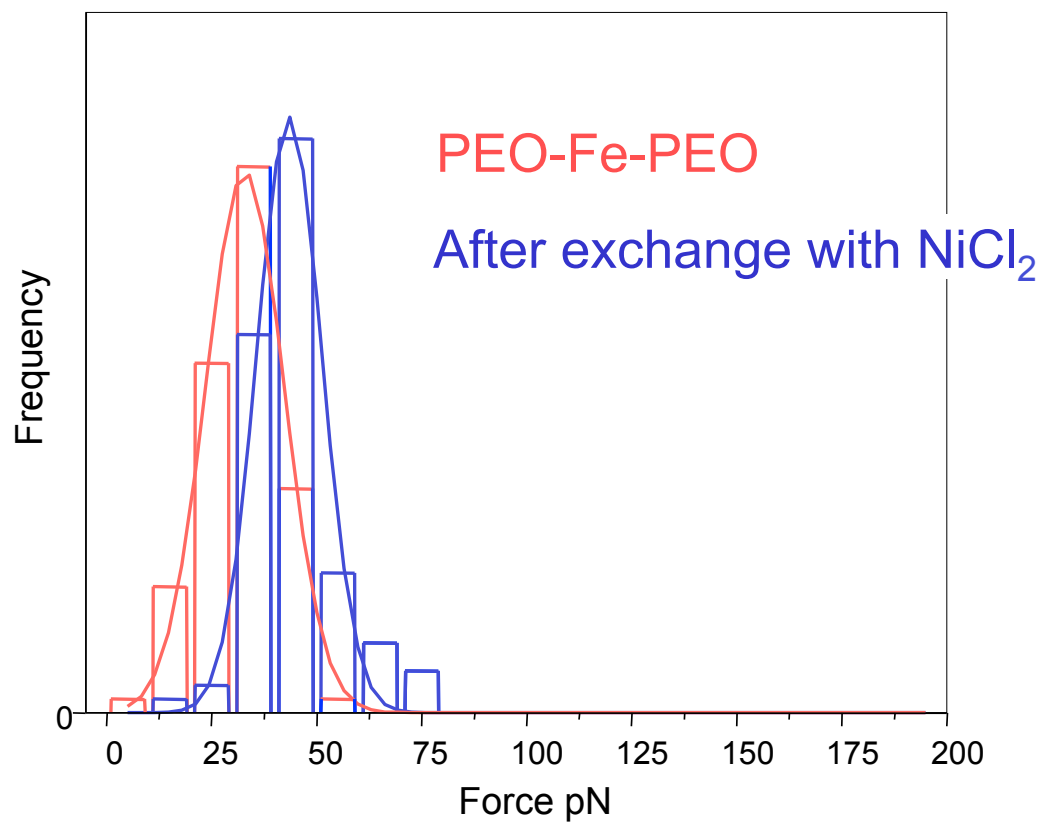
Terp-Co²⁺ 2.5 x 10⁸ M⁻¹

Terp-Fe²⁺ 1.3 x 10⁷ M⁻¹



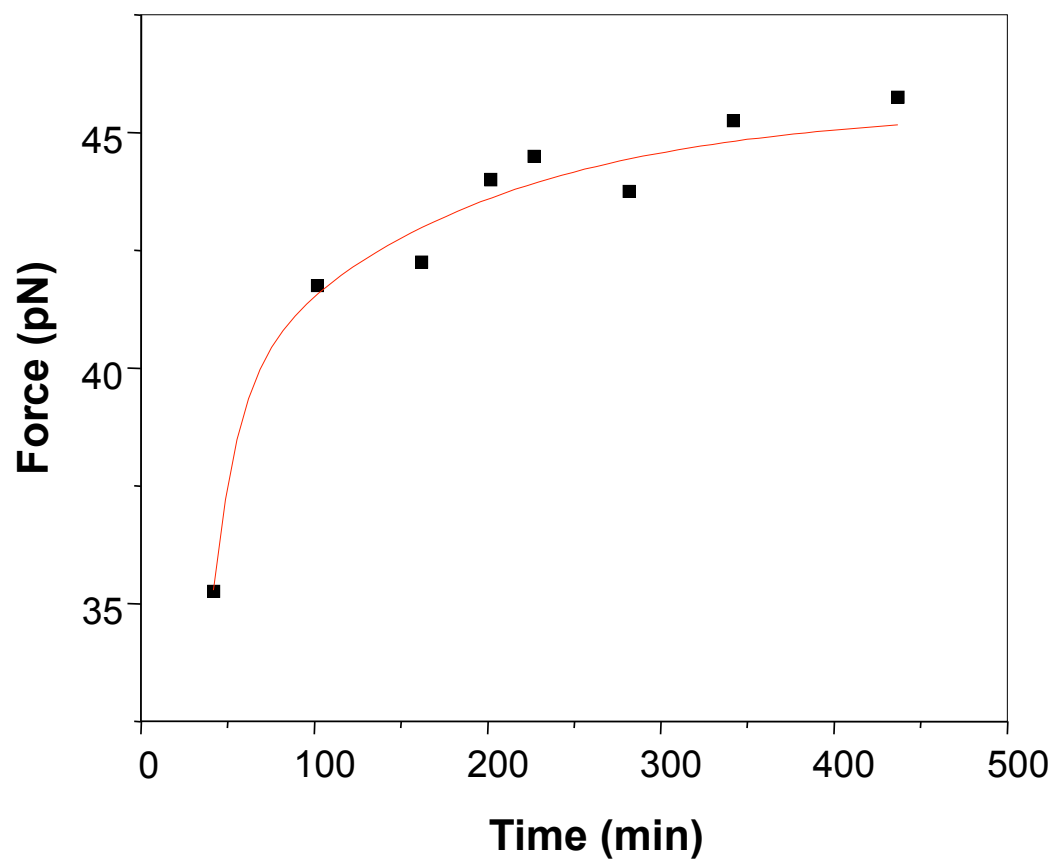
Metal exchange in a metallo-supramolecular chain

Exchange of Fe^{2+} with Ni^{2+}

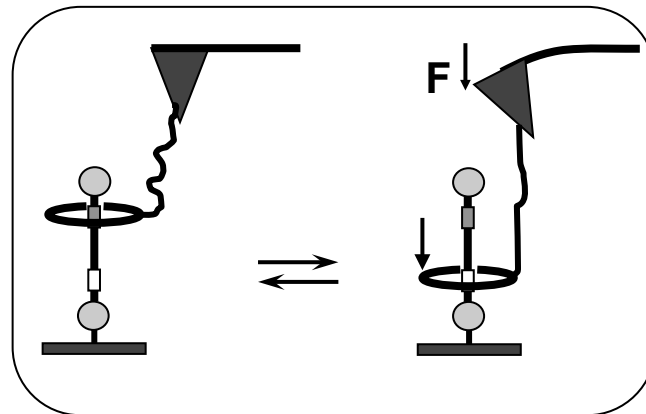


Metal exchange in metallo-supramolecular chains

Exchange of Fe^{2+} with Ni^{2+} : kinetics

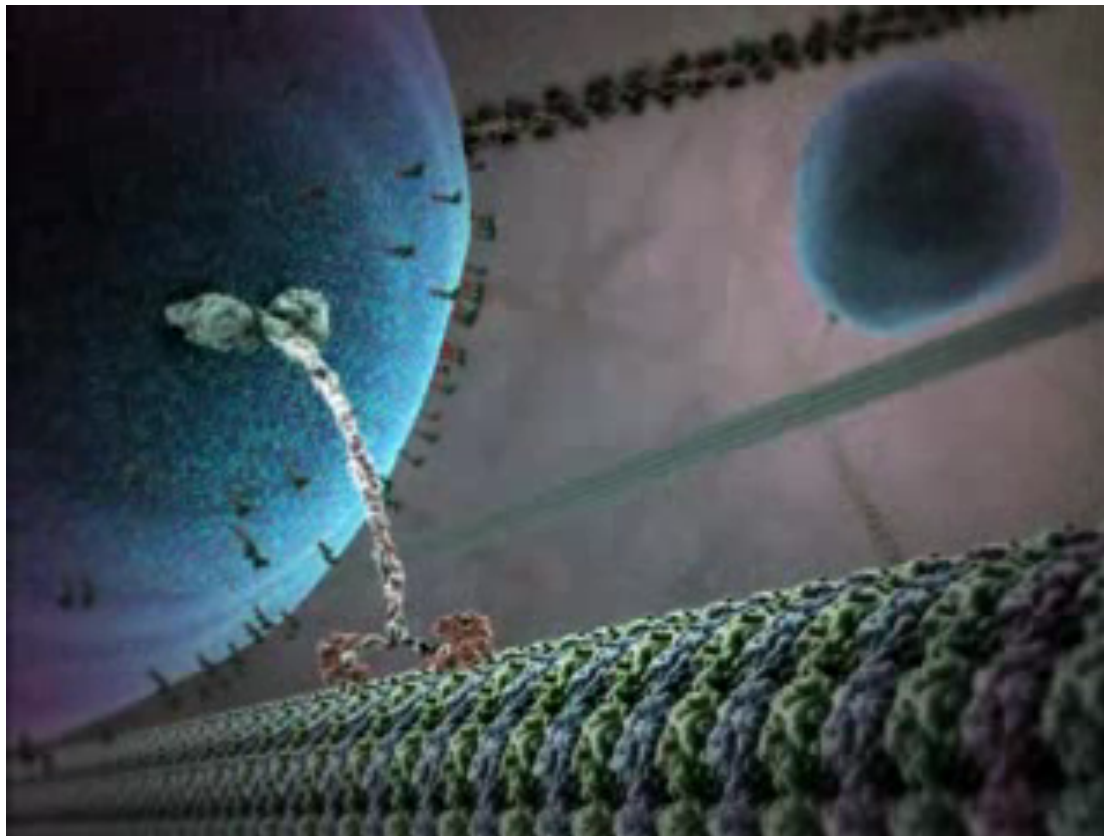


- Single molecule patterning
- Molecular interactions
- **Molecular machines**



Molecular machines are ubiquitous in nature and are essential in controlling and performing numerous biological functions

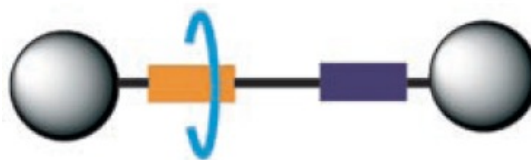
They are able to rectify random motion to generate directional force and carry out macroscopic tasks



'The Inner Life of the Cell' Biovisions, Harvard University
<http://multimedia.mcb.harvard.edu/>

➔ Inspiration for the design of synthetic systems able to use biased Brownian motion to perform work

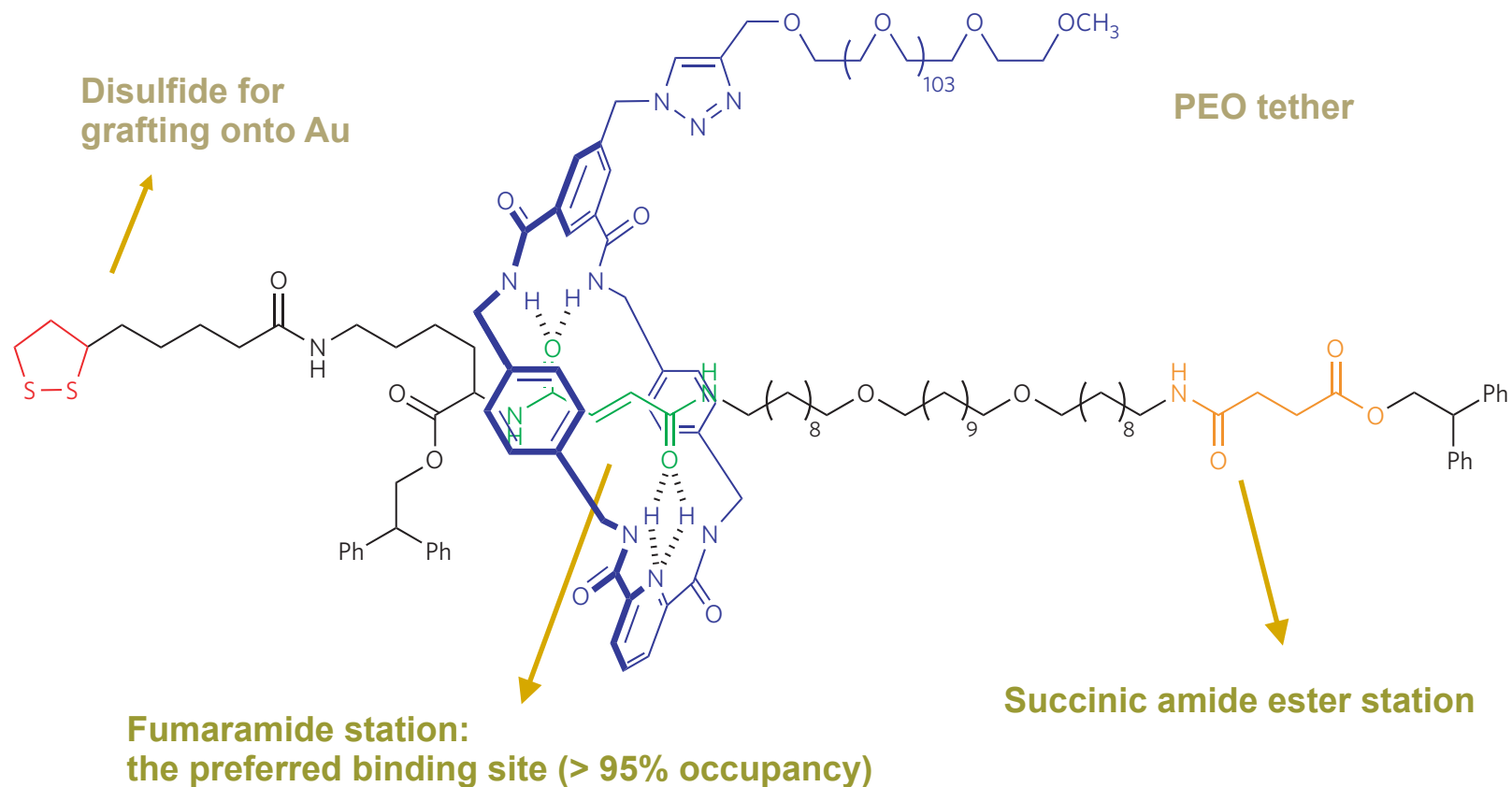
Rotaxane: a molecular ring threaded onto a molecular axle



Can sub-molecular Brownian motions in a **single** synthetic small machine be harnessed to generate forces ?

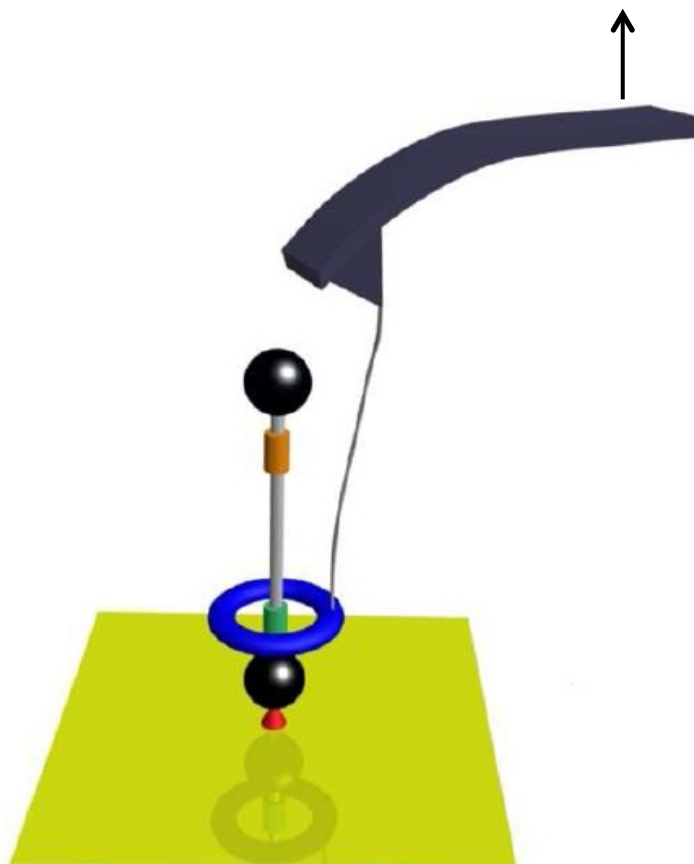
Can we measure the mechanical work done by the molecule?

Hydrogen-bonded rotaxane



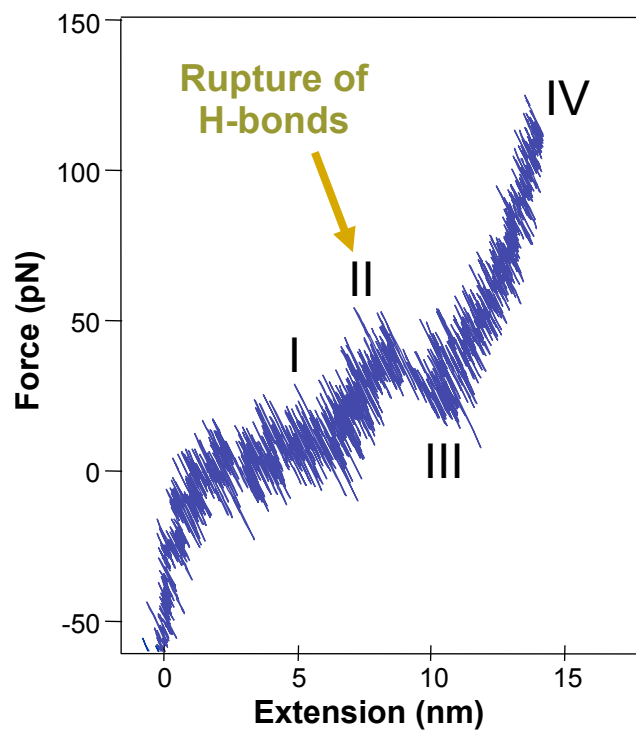
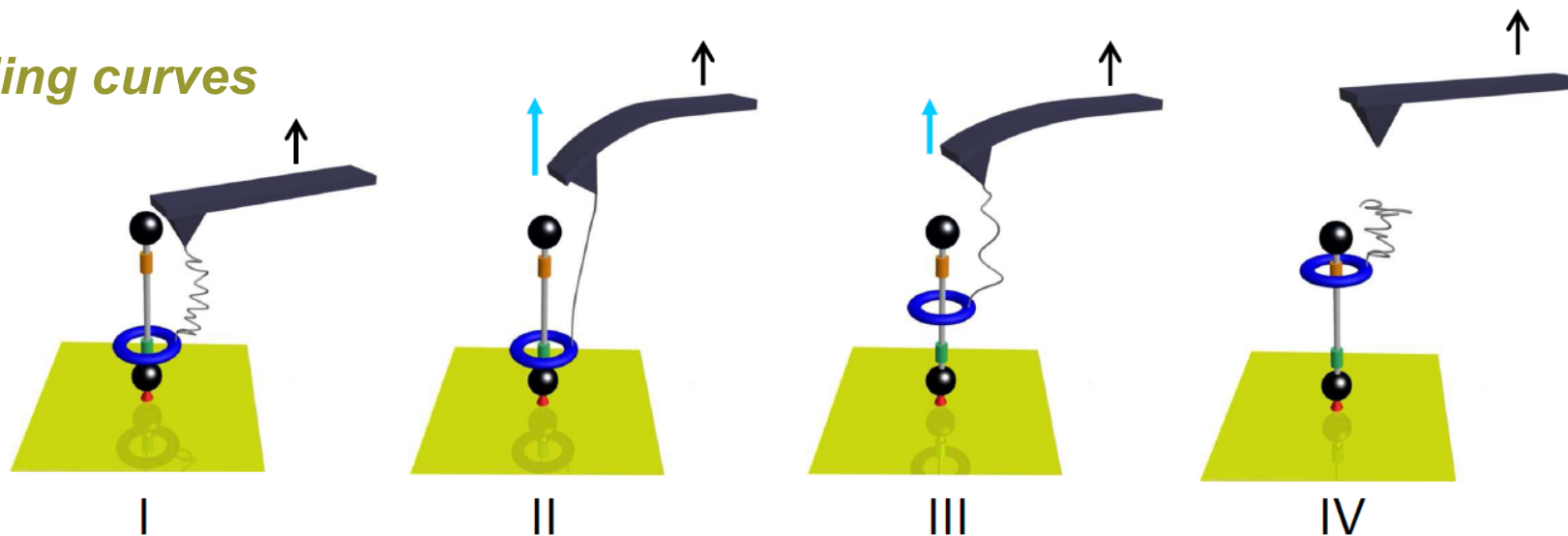
In collaboration with David A. Leigh, Univ. Manchester
and Charles-André Fustin, UCL

Interfacing molecular machines with the AFM

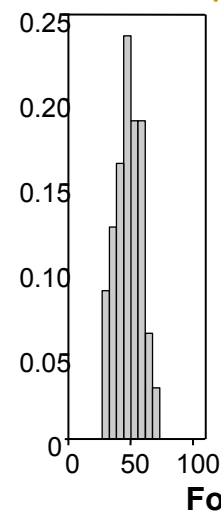


In collaboration with David A. Leigh, Univ. Manchester
and Charles-André Fustin, UCL

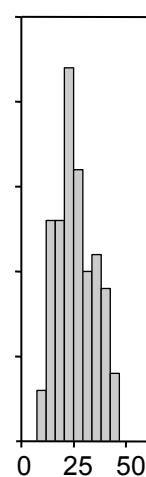
Pulling curves

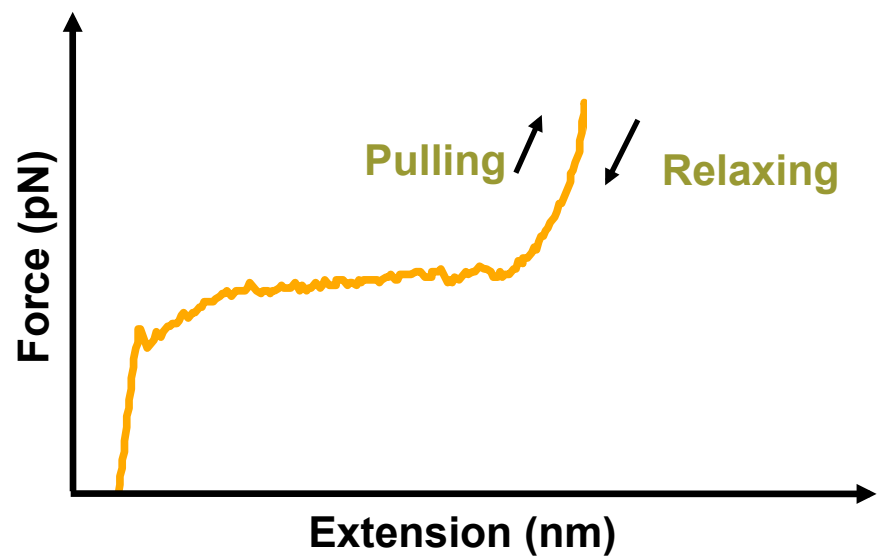
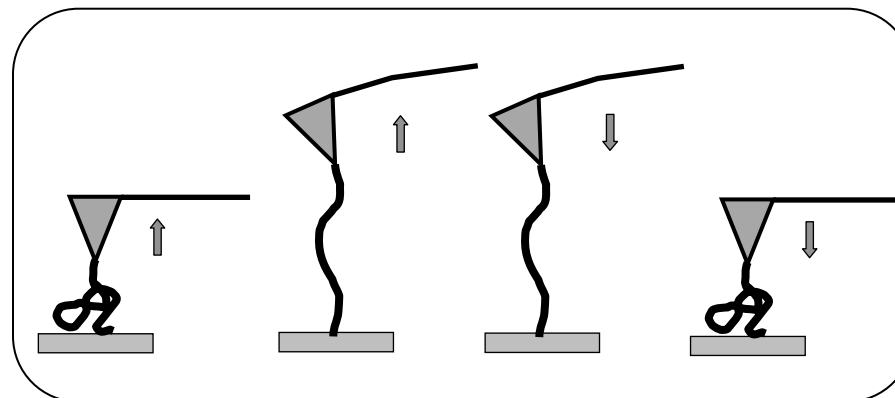
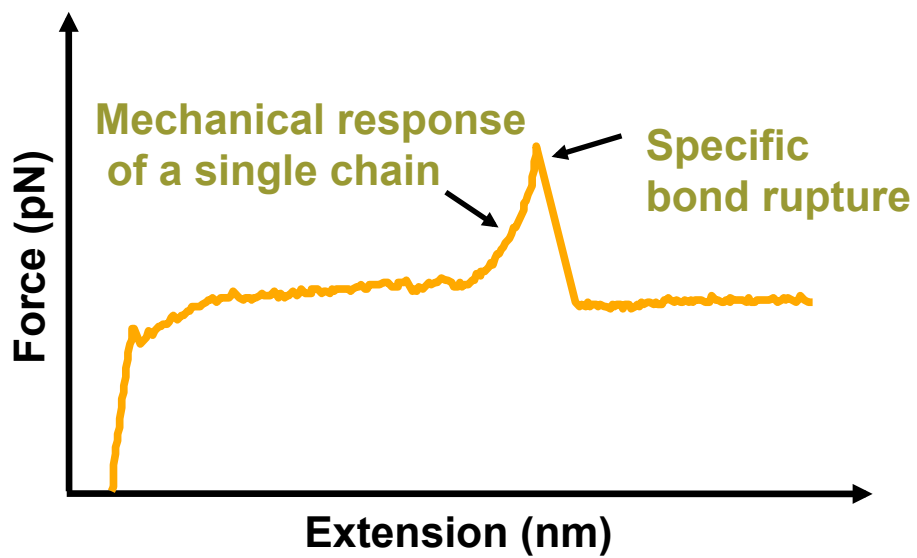
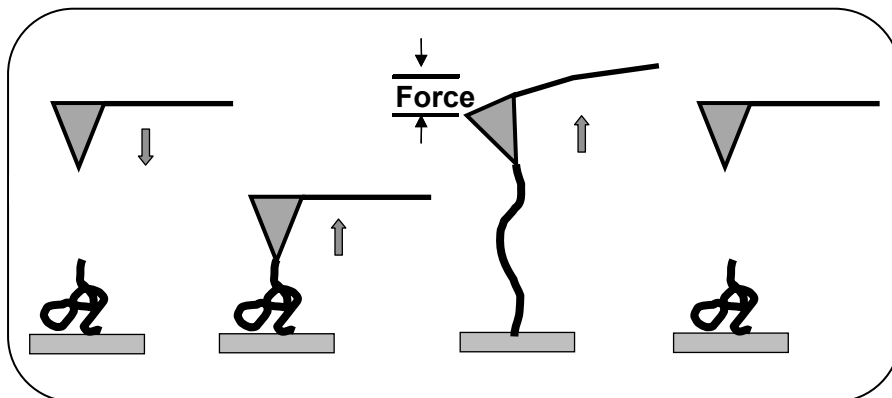


TCE
 45 ± 10 pN

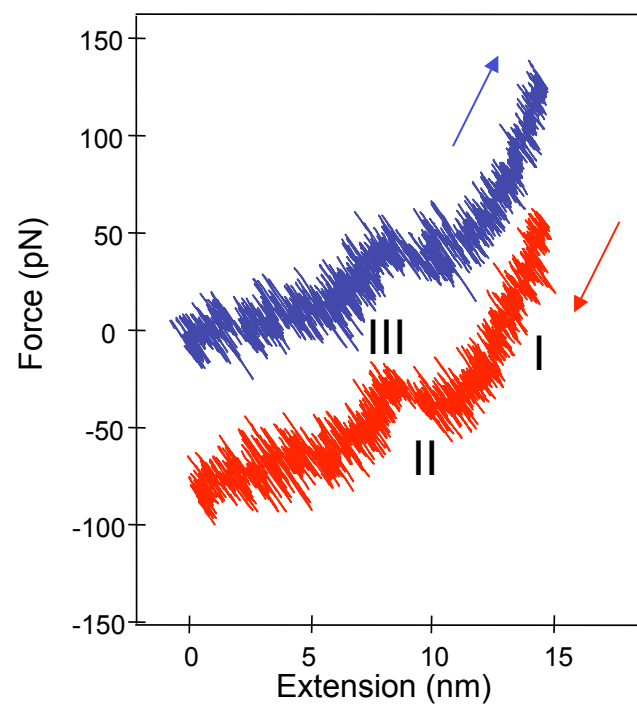
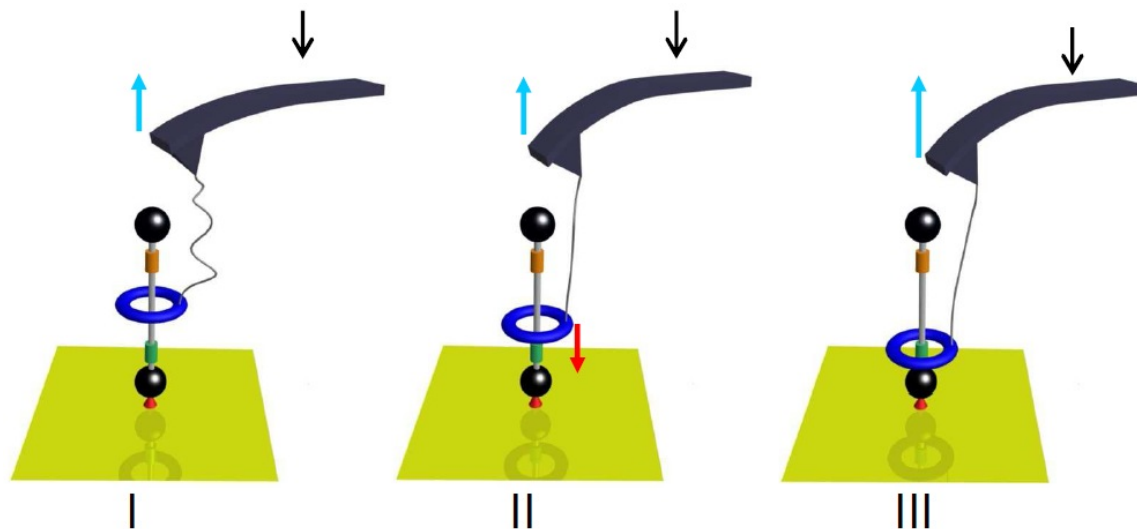


DMF
 27 ± 8 pN





Pulling-relaxing cycles



Force generation against
load

Shuttling against 30 pN

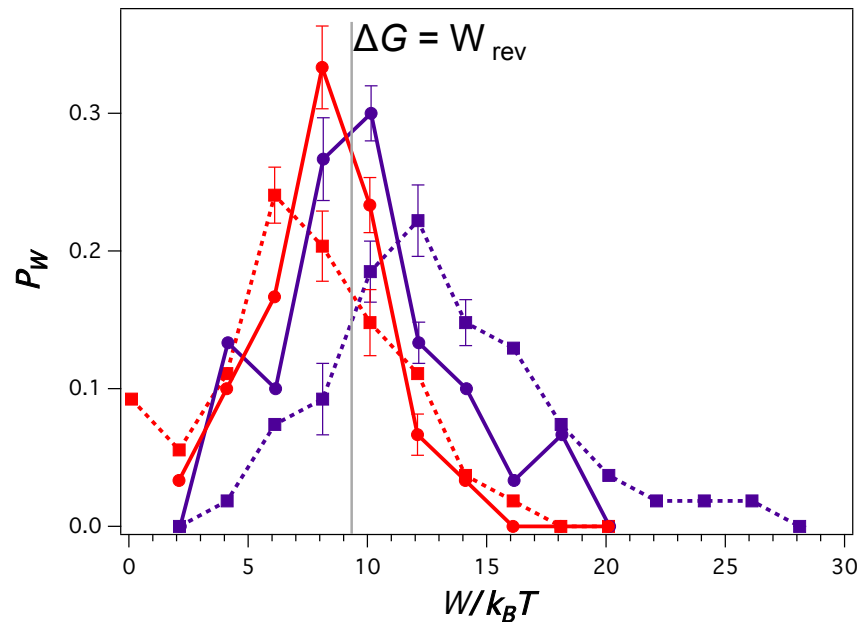
Work done by the ring:
 $\sim 6 \text{ kcal mole}^{-1}$

Nature Nanotech. **2011**, 6, 553

- Yes, sub-molecular Brownian motions in a single synthetic small machine can be harnessed to generate forces
- For a loading rate of 500 pN s^{-1} , the rotaxane is able to utilize almost all the energy available from hydrogen bonding to perform work along the direction of the applied load

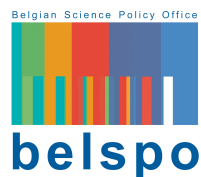
The principles of thermodynamics describe processes of energy exchange (work and heat) of macroscopic systems with their environment

Nonequilibrium systems are characterized by irreversible heat losses between the system and its environment, typically a thermal bath

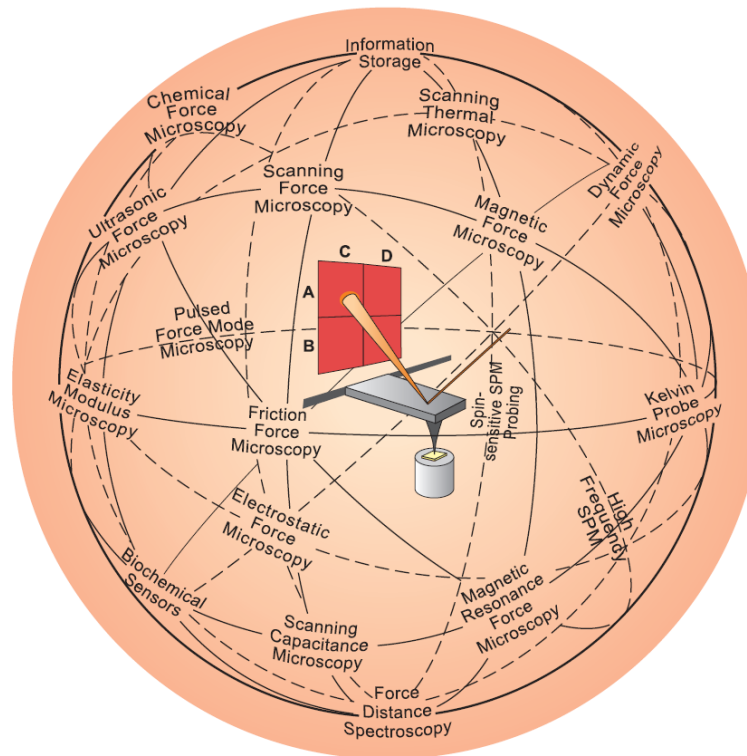


A single molecule can save heat loss to produce work

- ◆ PhD students and post-docs:
P. Lussis, N. Willet, N. Giamblanco, A. Burmistrova, F. Bano, A. Wislez, T. Svaldo-Lanero, M. Asano, C. Pain, D. Sluysmans, F. Devaux
- ◆ C. Jérôme, F. Remacle (University of Liège)
- ◆ C. A. Fustin (Université catholique de Louvain)
- ◆ D. A. Leigh (University of Manchester, UK)
- ◆ C. Jarzynski (University of Maryland, MD, USA)
- ◆ J. F. Stoddart (University of Northwestern, IL, USA)



MAY THE FORCE BE WITH YOU ...



From Ch. Gerber and H. P. Lang, *Nature Nanotech.* **2006**, 1, 3